



7) A, CsF, *n*-Bu₄NBr, MeCN, -20 °C
8) PDC, *t*-BuOOH, PhH
9) Cu(OAc)₂•H₂O, BDP (50 mol%), PMHS, *t*-BuOH, PhMe
10) MeCN, 80 °C *then* Cu(OTf)₂ (20 mol%), 40 °C
11) acryloyl chloride, TEA, DCM
12) Grubb's II (20 mol%), DCM, 40 °C
13) Rh(cod)(acac) (50 mol%), PhSiH₃, DCM, 40 °C
14) TFA, DCM, -78 °C to 23 °C

7) Rationalize the regiochemical outcome of this reaction. This reaction proceeds via an inverse electron-demand Diels-Alder (electron-poor diene, electron-rich dienophile), so the more substituted dienophile (the trisubstituted olefin of the allene) reacts. The pyrone carbonyl is thought to provide the dominant electronic effect, guiding bond formation between $C\alpha$ and the more substituted carbon of the cyclic allene:

